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A Four Component Biginelli's Reaction, New Opportunities for Synthesis of Functionalized Pyrimidines

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ABSTRACT

Methyl 5-aroyl-6-aryl-4-methoxy-2-oxo(thioxo)hexahydropyrimidine-4-carboxylates were synthesized by four-component Biginelli's reaction of methyl aroylpyruvate, aromatic aldehyde, urea (thiourea) and methanol in presence the sodium hydrogen sulfate . Products were obtained in moderate to high yields.

Keywords: Biginelli's Reaction; Methyl Aroylpyruvate; Aromatic Aldehyde; Urea; Thiourea; Sodium Hydrogen Sulfate; Methanol; Methyl 5-Aroyl-6-Aryl-4-Methoxy(Hydroxy)-2-Oxo(Thioxo)Hexahydropyrimidine-4-Carboxylates

Introduction

In 1893 year, three-component acid-catalyzed condensation of ethyl acetoacetate, benzaldehyde, and urea for synthesis of 3,4-dihydropyrimidin-2(¹H) ones (DHPMs) was reported by an Italian chemist Pietro Biginelli [1]. The great interest of chemists in Biginelli condensation provided access to many multi functionalized dihydropyrimidines due to the expansion of synthetic potentials of the well-known Biginelli reaction [2]. It is known that 5-functionally-substituted dihydropyrimidin-2(1H)ones (thiones) (5-carbonyl-, 5-alkoxycarbonyl-, 5-carbamoyl-, 5-carbonitryl-, 5-nitro-) have a wide range of biological activity (Figure 1). Among them

was found compounds with antitumor [3], anti-inflammatory [4], anti-tuberculosis [5], antimycobacterial [6], antioxidant [7], fungicidal [8], antihypertensive [9] activities. Some of them are used as calcium channel blockers [10]. The biological screening showed prospects of using of known N(3)-substituted 5-alkoxycarbonyldihydropyrimidin-2(1H)ones(thiones) as vasodilatators (3-alkoxycarbonyl-),[11] cardiovascular (3-alkyl-, 3-acyl-) [12] antimalarial [13] (3-carbamoyl-), anti-inflammatory (3-methyl-, phenazyl-)[14] agents and of antagonists of adrenergic receptors [15]. Among the N(1)-substituted 5-carbonyldihydropyrimidin-2(1H)ones(thiones) show carcinostatics (1-alkyl-) [16].

Figure 1.

The chemical modification of the pyrimidine ring, by replacing the original components of the classic Biginelli's reaction, can lead to strengthening or changing the pharmacological properties, reduce toxicity and given the practical significance of pyrimidine derivatives, chemists' interest in finding a new ways to modify of the pyrimidine cycle to produce new pyrimidine derivatives with potential biological activity is relevant. We previously studied the possibility of using methyl aroylpyruvates in reaction Biginelli with aromatic aldehydes, and urea in the absence of a solvent and a catalyst and received of methyl 6-aryl-5-aroyl-2-oxo-1,2,3,6-tetrahydropyrimidine-4-carboxylates as the only products [17]. Data on the synthesis of DHPMs with two substituents at the 4-position of the pyrimidine ring are scarce. Previously, the authors reported the formation of 4-hydroxytetrahydropyrimidinones in the reaction Biginelli if the structure of the dicarbonyl compound contain strong electron-acceptor substituents [18]. In preceding publication, we described the formation of 5-hydroxy-4,5,6,7-tetrahydrotetrazolo[1,5-a]pyrimidines in the reaction of methyl acetylpyruvate, aromatic aldehyde, and 5-aminotetrazole monohydrate [19]. We recently discovered and reported a simple way of obtaining non-dehydrated reaction products of Biginelli's reaction with simultaneous methylation of the tertiary hydroxyl group by reaction of methyl benzoylpyruvate with an aromatic aldehyde and urea in the presence of nontoxic, cheap sodium hydrogen sulfate as catalyst with formation three examples of methyl 5-benzoyl-6-aryl-4-me-

thoxy-2-oxohexahydropyrimidine-4-carboxylates. The reaction was carried out by heating in methanol [20].

Continuing research in this area, in this paper, we extended the range of aromatic aldehydes in the reaction with methyl aroylpyruvates and urea and investigated the possibility of replacing urea with thiourea in this reaction. It was found that the reaction similarly stops at the stage of formation of an undehydrated product with methylation of a tertiary hydroxyl group and as a result, we isolated new noncharacteristic products for Biginelli's reaction, in good yields (Scheme 1). 5: R= 4-HO C₆H₄ (a), 3-MeOC₆H₄ (b), 4-MeOC₆H₄ (c, g), 2,4-(MeO)₂ C₆H₂ (d), 4-Cl C₆H₄ (e, h), 4-F C₆H₄ (f, i), $(CH_2)2N C_6H_4$ (j); Ar = Ph (a-d, g-j), 4-CH₂ C_6H_4 (e, f), X = O (af), S (g-j). Despite the large number of publications on the study of the Biginelli's reaction, its mechanism is still under debate. Several mechanisms for the acid-catalyzed reaction have been reported. In this case a suggested mechanism is depicted in Scheme 2 and at the stage of the formation of 6-aryl-5-aroyl-4-hydroxy-2-oxohexahydropyrimidine-4-carboxylate a methylation occurs of the hemiaminal hydroxy group catalyzed by acidic by the nature sodium hydrogen sulfate. The electron-acceptor methoxycarbonyl and benzoyl groups in the positions 4 and 5 of the heterocycle stabilize structure A [21]. The reaction of obtaining the compound 5j stops at the stage of formation of structure A due to the basic properties of dimethylamino group of aromatic aldehyde, which possibly blocks the action of the catalyst.

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Scheme 1: Synthesis of methyl 5-aroyl-6-aryl-4-methoxy-2-oxo(thioxo)hexahydropyrimidine-4-carboxylates (5a-i).

Scheme 2: Proposed mechanism for the formation of methyl 5-aroyl-6-aryl-4-methoxy-2-oxo(thioxo)hexahydropyrimidine-4-carboxylates.

The structures were confirmed by ¹H NMR, ¹³C NMR and IR spectroscopy, as well as elemental analysis. Additionally, the structures of methyl 5-benzoyl-4-methoxy-6-(2,4-dimethoxy-phenyl)-2-oxohexahydropyrimidine-4-carboxylate (5d), methyl 5-benzoyl-6-(4-chlorophenyl)-4-methoxy-2-thioxohexahydropyrimidine-4-carboxylate (5h), methyl 5-benzoyl-6-(4-(dimethylamino)phenyl)-4-hydroxy-2-thioxohexahydropyrimidine-4-carboxylate (5j) were confirmed by single crystal X-ray analysis (Figures 2- 4). Substitution of methanol for ethanol in this reaction led to ethylation of the tertiary hydroxy group and partial transesterification of methoxycarbonyl group (Scheme 3). The structure of methyl(ethyl) 5-benzoyl-6-(3-bromophenyl)-4-ethoxy-2-oxohexahydropyrimidine-4-carboxylate were confirmed by ¹H NMR and single crystal X-ray analysis (Figure 5). CCDC 1833000 (for 5d), CCDC

1832999 (for 5h), CCDC 1832998 (for 5j) and CCDC 1832997 (for 6a), contain the supplementary crystallographic data for this paper. The data can be obtained free of charge from The Cambridge Crystallographic Data Centre via http://www.ccdc.cam.ac.uk. In summary, we have identified new possibilities for the Biginelli reaction, which allow us to introduce the methoxy group into the 4-position of the heterocycle by four-component Biginelli's reaction, which allows us to significantly expand the range of use of Biginelli reaction in the synthesis of biologically active compounds. For the first time shown the influence of the catalyst, the nature of the substituent in aromatic aldehyde and solvent on the structure of the obtained DHPMs. The obtained compounds can be used for the synthesis of condensed heterocyclic compounds (Figure 6).

Scheme 3: Synthesis of methyl(ethyl) 5-benzoyl-6-(3-bromophenyl)-4-ethoxy-2-oxohexahydropyrimidine-4-carboxylate (6a).

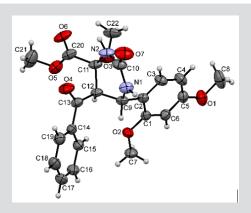


Figure 2: Molecular structure of compound 5d showing 50% probability amplitude displacement ellipsoids (CCDC 1833000).

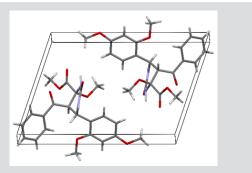


Figure 3: Packing of compound 5d in crystalline lattice.

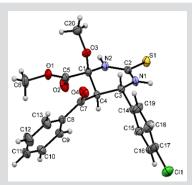


Figure 4: Molecular structure of compound 5h showing 50% probability amplitude displacement ellipsoids (CCDC 1832999).

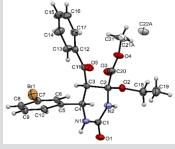


Figure 5: Molecular structure of compound 6a showing 50% probability amplitude displacement ellipsoids (CCDC 1832997). The minor disorder component is shown with white carbons.

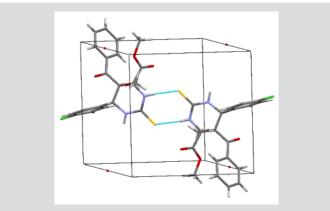


Figure 6: Centrosymmetric dimer in the crystal of compound 5h.

Experimental Section

All reagents and solvent were commercially available with analytical grade and used as received. Melting points were taken on a M-565. Infrared spectra (IR) were obtained on a Specord M-80 spectrometer in KBr. The ¹H NMR spectra were recorded on a Bruker 300 instruments using DMSO-d6 as solvent. The unit cell parameters and the X-ray diffraction intensities were measured on a Xcalibur Ruby diffractometer. The unit cell parameters and the X-ray diffraction intensities were measured on a Xcalibur Ruby diffractometer (Figure 7). The empirical absorption correction was introduced by multi-scan method using SCALE3 ABSPACK algorithm [21]. The structures were solved by charge flipping algorithm using the Superflip program [22] (5h, 5j) or olex [2]. solve program [23] (6a) or by direct method using the SHELXS-97 program [24] (5d) and refined by the full-matrix least-squares method in the anisotropic approximation for all non-hydrogen atoms using the SHELX-2014 [25] and OLEX2[26] program packages. Hydrogen atoms bound to carbon were located from the Fourier synthesis of the electron density and refined using a riding model. The hydrogen atoms of NH and OH groups were refined freely with isotropic displacement parameters. The H atoms of water molecule (5h) could not be located in the difference Fourier map or placed geometrically, and therefore were omitted from the refinement [27-32].

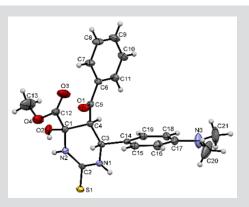


Figure 7: Molecular structure of compound 5j showing 50% probability amplitude displacement ellipsoids (CCDC 1832998).

Representative Procedure General Procedure for the Synthesis of Methyl 5-Aroyl-6-Aryl-4-Methoxy-2-Oxo(Thioxo) Hexahydropyrimidine-4-Carboxylates(5a-I): A mixture of methyl aroylpyruvate (1) (0.01 mol), aromatic aldehyde (2) (0.01 mol) and urea (thiourea) (3) (0.015 mol) and MeOH (10 mL) (4) in presence the NaHSO4 (0.008 mol) in was heated at for 30 min⁻¹ h. After reaction completion, the resulting precipitate was collected by filtration and washed with water, the crude product was recrystallized from MeOH (Figure 8).

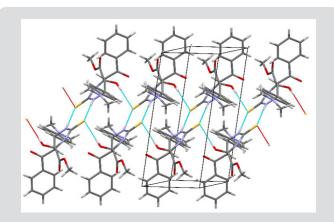


Figure 8: One-dimensional double chain in the crystal of compound 5j.

The Procedure for the Synthesis of Methyl(ethyl) 5-Benzoyl-6-(3-Bromophenyl)-4-Ethoxy-2-Oxohexahydropyrimidine-4-Carboxylate (6a): A mixture of methyl benzoylpyruvate (1) (0.01 mol), 3-bromobenzaldehyde (2) (0.01 mol) and urea (3) (0.015 mol) and EtOH (10 mL) (4) in presence the NaHSO4 (0.008 mol) in was heated at for 1 h. After reaction completion, the resulting precipitate was collected by filtration and washed with water, the crude product was recrystallized from EtOH.

Methyl5-Benzoyl-6-(4-Hydroxyphenyl)-4-Methoxy-2-Oxohexahydropyrimidine-4-Carboxylate (5a): Yield: 2.46 g (64%), colorless crystalline solid, mp 252-256°C; ¹H NMR (300 MHz, DM-SO-d6) δ, ppm (J, Hz): 3.09 (³H, s, COOCH₃); 3.19 (³H, s, CH₃O); 4.25 (¹H, d, J = 11 Hz, H-5); 4.91 (¹H, d, J = 11 Hz, H-6); 6.59 (²H, d, J = 8.4 Hz, ArH); 6.72 (¹H, s, 1-H); 7.19 (²H, d, J = 8.4 Hz, ArH); 7.39 (²H, t, J = 8.4 Hz, ArH); 7.52 (1H, t, J = 8.4 Hz, ArH); 7.59 (2H, d, J = 8.4 Hz, ArH); 7.76 (1H, s, 3-H); 9.14 (1H, s, OH) ppm; Found, %: C, 62.61; H, 5.33; N, 7.16. Anal. Calcd for C20H20N2O6: C, 62.49; H, 5.24; N, 7.29%.

Methyl-5-Benzoyl-6-(3-Methoxyphenyl)-4-Methoxy-2-Oxohexahydropyrimidine-4-Carboxylate (5b): Yield: 2.70 g (68%), colorless crystalline solid, mp 162-164°C. IR (KBr) ν (cm $^{-1}$) 1664, 1724, 3168, 3354. 1 H NMR (300 MHz, DMSO-d6) δ, ppm (J, Hz): 3.09 (3 H, s, COOCH $_{3}$); 3.18 (3 H, s, CH3O); 3.70 (3 H, s, CH3O); 4.39 (1 H, d, J = 11 Hz, H-5); 5.03 (1 H, d, J = 11 Hz, H-6); 6.74 (1 H, d, 1-H); 6.98 (1 H, s, ArH); 7.00 (2 H, d, J = 7.8 Hz, ArH); 7.15 (1 H, t, J = 7.5 Hz, ArH); 7.42 (2 H, t, J = 7.8 Hz, ArH); 7.55 (1 H, t, J = 7.2 Hz, ArH); 7.67 (2 H, d, J = 7.2 Hz, ArH); 7.95 (1 H, s, 3-H) ppm. Found, %: C, 63.45; H, 5.66; N, 7.14. Anal. Calcd for C21H22N2O6: C, 63.31; H, 5.57; N, 7.03%.

Methyl5-Benzoyl-4-Methoxy-6-(4-Methoxyphenyl)-2-Oxohexahydropyrimidine-4-Carboxylate (5c): Yield: 2.35 g (59%), colorless crystalline solid, mp 224-226°C. 1 H NMR (300 MHz, DM-SO-d6) δ, ppm (J, Hz): 3.10 (3 H, s, COOCH $_3$); 3.19 (3 H, s, CH $_3$ O); 4.28 (4 H, d, J = 11 Hz, H-5), 4.98 (4 H, d, J = 11 Hz, H-6); 6.70 (4 H, s, 1-H); 6.78 (2 H, d, J = 8.7 Hz, ArH); 7.32 (2 H, d, J = 8.7 Hz, ArH); 7.39 (2 H, t, J = 7.2 Hz, ArH); 7.52 (4 H, t, J = 7.2 Hz, ArH); 7.62 (2 H, d, J = 7.2 Hz, ArH); 7.71 (4 H, s, 3-H) ppm. 13C NMR (75 MHz, DMSO-d6) δ, ppm: 50.59; 51.65; 51.73; 53.05; 54.88; 85.54; 113.47; 127.46; 128.39; 129.18, 131.67; 132.85; 137.56; 153.19; 158.72; 167.73; 194.20. Found, %: C, 63.43; H, 5.48; N, 7.16. Anal. Calcd for $C_{21}H_{22}N_2O_6$: C, 63.31; H, 5.57; N, 7.03%.

Methyl5-Benzoyl-4-Methoxy-6-(2,4-Dimethoxy-phenyl)-2-Oxohexahydropyrimidine-4-Carboxylate (5d): Yield: 2.65 g (62%), colorless crystalline solid, mp 209-211°C. 1 H NMR (300 MHz, DMSO-d6) δ, ppm (J, Hz): 2.79 (3 H, s, COOCH3); 3.50 (3 H, s, CH3O); 3.75 (3 H, s, CH $_{3}$ O); 3.87 (3 H, s, CH3O); 4.48 (1 H, br s, H-5); 4.76 (1 H, br s, H-6); 6.51 (1 H, d, J = 8.7 Hz, ArH); 6.56 (1 H, s, ArH); 6.69 (1 H, s, 1-H); 6.93 (1 H, s, 3-H); 7.06 (1 H, d, J = 8.4 Hz, ArH); 7.61 (2 H, t, J = 6.9 Hz, ArH); 7.70 (1 H, t, J = 7.2 Hz, ArH); 8.04 (2 H, d, J = 7.2 Hz, ArH). Found, %: C, 61.80; H, 5.40; N, 7.05. Anal. Calcd for C $_{22}$ H $_{24}$ N $_{2}$ O $_{7}$: C, 61.68; H, 5.65; N, 6.54%.

Crystal Data of 5d. $C_{22}H_{24}N_2O_7$, M=428.43, triclinic, a=7.3891(9) Å, b=10.4223(12) Å, c=15.1407(17) Å, $\alpha=69.918(10)$ °, $\beta=81.948(10)$ °, $\gamma=69.871(11)$ °, $\gamma=1027.9(2)$ Å3, $\gamma=295(2)$, space group P–1, $\gamma=2$ 0, $\gamma=1027.9(2)$ Å3, $\gamma=295(2)$ 0, space group P–1, $\gamma=2$ 1, $\gamma=2$ 2, $\gamma=2$ 3, $\gamma=$

Methyl 6-(4-Chlorophenyl)-4-Methoxy-5-(4-Methylbenzoyl)-2-Oxohexahydropyrimidine-4-Carboxylate (5e): Yield: 2.96 g (71%), colorless crystalline solid, mp 218-220°C. 1 H NMR (300 MHz, DMSO-d6) δ, ppm (J, Hz): 2.30 (3 H, s, CH $_{3}$); 3.14 (3 H, s, C00CH $_{3}$);3.19 (3 H, s, CH $_{3}$ O); 4.28 (1 H, d, J = 11 Hz, H-5); 5.00 (1 H, d, J = 11 Hz, H-6); 6.83 (1 H, s, 1-H); 7.21 (2 H, d, J = 8.4 Hz, ArH); 7.26 (2 H, d, J = 8.7 Hz, ArH); 7.42 (2 H, d, J = 8.7 Hz, ArH); 7.54 (2 H, d, J = 8.4 Hz, ArH); 7.73 (1 H, s, 3-H) ppm. 13 C NMR (75 MHz, DMSO-d6) δ, ppm: 20.88; 50.64; 51.32; 51.86; 53.45; 85.60; 127.65; 127.97; 128.37; 128.85; 129.83; 132.11; 138.85; 142.94; 153.18; 167.64; 193.48. Found, %: C, 60.38; H, 5.17; N, 6.60.Anal.Calcd for C $_{21}$ H $_{21}$ ClN $_{2}$ O $_{5}$: C, 60.51; H, 5.08; N, 6.72%.

Methyl 6-(4-Fluorophenyl)-4-Methoxy-5-(4-Methylbenzoyl)-2-Oxohexahydropyrimidine-4-Carboxylate (5f): Yield: 2.96 g (74%), colorless crystalline solid, mp 226-228°C. 1 H NMR (300 MHz, DMSO-d6) δ, ppm (J, Hz): 2.29 (s, 3 H, CH $_3$); 3.14 (3 H, s, COOCH $_3$); 3.19 (3 H, s, CH $_3$ O); 4.28 (1 H, d, J = 11 Hz, H-5); 5.00 (1 H, d, J = 11 Hz, H-6); 6.82 (1 H, s, 1-H); 7.00 (2 H, t, J = 8.7 Hz, ArH); 7.20 (2 H, d, J = 8.1 Hz, ArH); 7.40-7.45 (2H, m, ArH); 7.54 (2H, d, J = 8.1 Hz,

ArH); 7.74 (s,1H, 3-H) ppm. 13C NMR (75 MHz, DMSO-d6) δ, ppm: 20.74; 50.52; 51.45;51.73; 53.03; 85.54; 114.47; 114.75; 127.49; 128.82; 129.91; 130.02; 135.05; 135.87; 135.97;143.29; 153.09; 159.81; 163.04; 167.59; 193.42. Found, %: C, 62.87; H, 5.37; N, 7.11. Anal. Calcd for C₂₁H₂₁FN₂O₂: C, 62.99; H, 5.29; N, 7.00% (Figure 9).

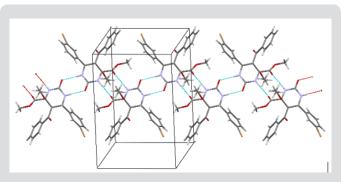


Figure 9: One-dimensional chain in the crystal of compound 6a.

Methyl 5-Benzoyl-4-Methoxy-6-(4-Methoxyphenyl)-2-Thioxohexahydropyrimidine-4-Carboxylate (5g): Yield: 2.81 g (68%), colorless crystalline solid, mp 206-208°C. 1 H NMR (300 MHz, DMSO-d6) δ, ppm (J, Hz): 3.14 (3 H, s, COOCH $_3$); 3.22 (3 H, s, CH $_3$ O); 3.66 (3 H, s, CH $_3$ O); 4.38 (4 H, d, J = 11 Hz, H-5); 4.93 (4 H, d, J = 11 Hz, H-6); 6.81 (2 H, d, J = 8.7 Hz, ArH); 7.33 (2 H, d, J = 8.7 Hz, ArH); 7.39 (2 H, t, J = 7.5 Hz, ArH); 7.54 (4 H, t, J = 7.5 Hz, ArH); 7.63 (2 H, d, J = 7.2 Hz, ArH); 8.48 (4 H, s, 1-H); 9.20 (4 H, s, 3-H) ppm. 13 C NMR (75 MHz, DMSO-d6) δ, ppm: 49.88; 51.38; 54.01; 54.91; 51.92; 84.11; 113.57; 127.63; 128.3; 129.42; 130.16; 133.12; 137.21; 158.93; 166.84; 175.86; 193.77. Found, %: C, 60.98; H, 5.25; N, 6.65. Anal. Calcd for C_{21} H $_{22}$ N $_2$ O $_5$ S: C, 60.85; H, 5.35; N, 6.76%.

Methyl 5-Benzoyl-6-(4-Chlorophenyl)-4-Methoxy-2-Thioxohexahydropyrimidine-4-Carboxylate (5h): Yield: 3.14 g (75%), colorless crystalline solid, mp 238-240°C. IR (KBr) v (cm-1): 1684, 1731, 3169, 3379. ¹H NMR (300 MHz, DMSO-d6) δ, ppm (J, Hz): 3.14 $(^{3}H, s, COOCH_{3}); 3.23 (^{3}H, s, CH_{3}O); 4.42 (^{1}H, d, J = 11 Hz, H-5); 4.99$ (1 H, d, J = 11 Hz, H-6); 7.30 (2 H, d, J = 8.4 Hz, ArH); 7.37 (2 H, d, J = 8.4 Hz, ArH); 7.43 (2 H, t, J = 8.4 Hz, ArH); 7.55 (1 H, t, J = 7.2 Hz, ArH); 7.66 (2 H, d, J = 7.2 Hz, ArH); 8.75 (1 H, s, 1-H); 9.26 (1 H, s, 3-H) ppm. ¹³C NMR (75 MHz, DMSO-d6) δ, ppm: 50.40; 50.80; 52.04; 53.96; 80.48; 127.74; 128.06; 128.35; 130.16; 133.12; 137.27; 137.44; 168.59; 175.82; 195.25. Found, %: C, 57.23; H, 4.48; N, 6.56. Anal. Calcd for C20H19ClN2O4S: C, 57.35; H, 4.57; N, 6.69%. Crystal Data of 5h. $2C_{20}H_{19}CIN_2O_4S\cdot O$, M = 853.76, triclinic, a = 8.0431(14) Å, b = 10.3361(18) Å, c = 12.506(2) Å, $\alpha = 88.241(14)$ °, $\beta = 86.090(14)$ °, $\gamma = 79.319(15)$ °, V = 1019.2(3) Å3, T = 295(2), space group P-1, Z = 1, μ(Mo Kα) = 0.321 mm⁻¹. The final refinement parameters: R1 = 0.0610, wR2 = 0.1520 [for observed 3057 reflections with I $> 2\sigma(I)$; R1 = 0.0990, wR2 = 0.1864 (for all independent 4713 reflections, Rint = 0.0441), S = 1.042. Largest diff. peak and hole 0.446 and -0.368 ēÅ-3.

Methyl 5-Benzoyl-6-(4-Fluorophenyl)-4-Methoxy-2-Thioxohexahydropyrimidine-4-Carboxylate (5i): Yield: 3.13 g (78%), colorless crystalline solid, mp 288-290°C. 1 H NMR (300 MHz, DMSO-d6) δ, ppm (J, Hz): 3.14 (s, 3 H, COOCH $_{3}$); 3.23 (s, 3 H, CH $_{3}$ O); 4.42 (1 H, d, J = 11 Hz, H-5); 4.99 (1 H, d, J = 11 Hz, H-6); 7.04 (2 H, t, J = 8.7 Hz, ArH); 7.37-7.48 (4 H, m, ArH); 7.54 (1 H, t, J = 7.2 Hz, ArH); 7.64 (2 H, d, J = 7.2 Hz, ArH); 9.24 (1 H, s, 3-H); 8.71 (1 H, s, 1-H) ppm. 13 C NMR (75 MHz, DMSO-d6) δ, ppm: 50.03; 51.39; 51.90; 54.01; 84.12; 114.72; 115.00; 127.61; 128.41; 130.30; 130.41; 133.13; 134.40; 134.44; 137.22; 160.06; 163.29; 166.74; 176.14; 193.76. Found, %: C, 59.84; H, 4.84; N, 6.83.Anal.Calcd for $C_{20}H_{19}FN_{2}O_{4}S$: C, 59.69; H, 4.76; N, 6.96%.

Methyl5-Benzoyl-6-(4-(Dimethylamino)phenyl)-4-Hydroxy-2-ioxohexahydropyrimidine-4-Carboxylate (5j): Yield: 1.61 g (39%), colorless crystalline solid, mp 140-142°C. ¹H NMR (300 MHz, DMSO-d6) δ, ppm (J, Hz): 2.81 (⁶H, s, N(CH₂)₂); 3.20 (³H, s, COOCH₂); 4.35 (¹H, d, J = 11 Hz, H-5); 4.89 (¹H, d, J = 11 Hz, H-6); 6.61 (2 H, d, J = 8.7 Hz, ArH); 6.88 (1 H, s, OH); 7.18 (2 H, d, J = 8.7 Hz, ArH); 7.38 (2 H, t, J = 7.5 Hz, ArH); 7.51 (1 H, t, J = 7.5 Hz, ArH); 7.64 (2 H, d, J = 7.5 Hz, ArH); 8.22 (1 H, s, 1-H); 8.61 (1 H, s, 3-H) ppm. Found: C, 61.12; H, 5.70; N, 10.29%. Anal. Calcd for C₂₁H₂₂N₂O₄S: C, 61.00; H, 5.61; N, 10.16%. Crystal Data of 5j. $C_{21}H_{23}N_3O_4S$, M = 413,48, triclinic, a = 5.7157(15) Å, b = 12.854(3) Å, c = 15.156(2) Å, $\alpha = 105.660(15)^{\circ}$, $\beta = 96.970(17)^{\circ}$, $\gamma = 99.642(19)^{\circ}$, $V = 1040.6(4)^{\circ}$ Å3, T = 295(2), space group P-1, Z = 2, μ (Mo K α) = 0.188 mm-1. The final refinement parameters: R1 = 0.0573, wR2 = 0.1503 [for observed 4382 reflections with I > $2\sigma(I)$]; R1 = 0.0863, wR2 = 0.1597 (for all independent 6732 reflections), S = 0.969. Largest diff. peak and hole 0.388 and -0.302 eÅ⁻³. The crystal is a nonmerohedral twin, the refined ratio of the twin components is 0.5635(14):0.4365(14).

Methyl (ethyl) 5-benzoyl-6-(3-bromophenyl)-4-ethoxy-2-oxohexahydropyrimidine-4-Carboxylate (6a): Yield: 2.91 g (63%), colorless crystalline solid, mp 198-200°C. ¹H NMR (300 MHz, DMSO-d6) δ , ppm (J, Hz): 0.86 (3 H, t, COOC₂H₂); 1.09 (3 H, t, OC₂H₂); 3.32 (²H, q, OC2H5); 3.53 (²H, q, COOC₂H₂); 3.04 (³H, s, COOCH₂); 4.37 (1H, d, J = 11 Hz, H-5); 5.04 (1H, d, J = 11 Hz, H-6); 7.10 (1H, s, 1-H); 7.19 (1 H, t, J = 7.8 Hz, ArH); 7.38 (2 H, d, J = 7.5 Hz, ArH); 7.44 (2 H, t, J = 7.8 Hz, ArH); 7.55 (¹H, t, J = 7.2 Hz, ArH); 7.65 (¹H, s, ArH);7.71 (²H, d, J = 7.2 Hz, ArH); 8.00 (¹H, s, 3-H) ppm; C_{21} .33H₂₁.66BrN₂O₅. Crystal Data of 6a. C_{21} .33 H_{21} .66 BrN_2O_5 , M = 465.94, monoclinic, a = 10.890(2) Å, b = 16.371(4) Å, c = 11.951(3) Å, β = 94.556(18) °, V = 2123.9(8) Å3, T = 295(2), space group P21/n, Z = 4, μ (Mo K α) = 1.970 mm⁻¹. The final refinement parameters: R1 = 0.0524, wR2 = 0.1070 [for observed 3241 reflections with I > $2\sigma(I)$]; R1 = 0.1298, wR2 = 0.1385 (for all independent 4916 reflections, Rint = 0.0387), S = 1.035. Largest diff. peak and hole 0.491 and $-0.935 \,\bar{e} \, \text{Å}^{-3}$.

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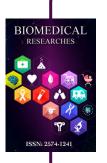
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