

Fraudulent Methods Causing False Negatives In Urine Drug Testing

Soycan Mizrak*

Clinical Biochemistry Laboratory, Usak Training and Research Hospital, Turkey

*Corresponding author: Soycan Mizrak, Clinical Biochemistry Laboratory, Usak Training and Research Hospital, Turkey



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ABSTRACT

Substance analyzes are carried out to determine drug use. There is a workflow plan that should apply for drug use tests carried out for medico-legal purposes. The most important thing about this is the preservation of urinary integrity. These tests mostly perform for security-sensitive professions or forensic situations. Therefore, drug users try fraudulent ways to hide their dependencies. Easily obtained chemicals are used for his purpose. Examples include nitrite, some commercial products such as peroxide, or household chemicals such as table salt or bleach. Always be aware of adulterants use. Proper monitoring during urine collection will greatly prevent fraudulent conditions. Another important point is to conduct urine integrity tests (pH, nitrite, density, and creatinine). Samples which urine integrity test (pH, nitrite, density, and creatinine) results in an acceptable range should be analyzed. We use immunoassay screening and GC MS validation methods for analysis. Immunoassay screening tests are affected by these adulterants than false negative results obtain due to the deteriorate antigen-antibody relationship. GC-MS verification tests should perform in case of clinical suspicion.

Mini Review

Fraudulent Methods Causing False Negatives in Urine Drug Testing

The urine drug test has a significant role in monitoring the use of legal and illicit substance [1]. Drug testing programs implement for all security-sensitive professions such as military forces, transport industry, the mining industry. Additionally, health care (facilitate treatment of clinical intoxication and rehabilitation programs), detect doping in sports, legal and criminal situations such as post-accident testing, rehabilitation testing of ex-convicts, are common areas for drug testing [2]. Drug testing screening programs are increasing day by day. Urine, blood, hair, saliva, sweat and nail samples are using for substance abuse screening tests. Depending on the method of measuring drug levels, screening tests have different specificity, sensitivity, and accuracy of each sample. The most frequently preferred sample for substance abuse screening test is urine because of its easy collection. Furthermore, urine has longer detection times and higher drug concentrations than blood [3]. At the stage of giving the urine sample can be easily cheated. Illegal drug users try to hide their substance use with adulterants

(*in vivo* or *in vitro*). *In vivo* adulteration involves the dilution of urine with water or taking a diuretic substance for increasing the metabolism.

In vitro adulteration is the act of adding foreign chemicals (such as vinegar, table salt, hypochlorite bleach, laundry detergent) to urine samples. Therefore, in order to prevent false negative results, the urine validity tests are used [4]. Studies have been carried out for many years to prevent false negative results by detecting adulteration action to the urine. For this purpose, urine specimen intake and admission procedures were developed. Surveillance of urine collection is the most important step. If we make this step meaningfully, we prevent urinary substitution and *in vitro* adulteration. The other steps of urine integrity tests are temperature, specific gravity, pH, nitrite and creatinine. measurements. Urinary sequencing DNA is the highest accuracy method of urine validation tests [5]. The cross-reaction of a urine adulterant with immunoassay tests include forming insoluble drug analyte-adulterant complexes. In this paper, I describe the oxidizing and non-oxidizing chemical adulterants which cross-react with the immunoassay method.

Oxidizing Chemicals

Nitrite: Nitrite in urine is caused by nitrate reducing pathogens in urinary tract infections. However, the commercial products KLEAR and Whizzies can purchase from markets and they contain potassium nitrite (KNO_2) and sodium nitrite (NaNO_2), respectively [6]. The advantage of nitrite adulterant is that it is easy to use and does not cause any changes in urine appearance. Nitrite analysis can be performed semiquantitatively by urine test strips or quantitatively by automated colorimetric methods. According to the increasing level of nitrite in urine when kept in room temperature for 6 or 8 days, it can be differentiated from the externally added commercial nitrite [7]. Nitrite ion can cause inconsistent results especially for cannabis testing, between the immunoassay screening and gas chromatographic-mass spectrometric (GC-MS) confirmation in urine. False negative immunoassay and GC-MS results obtain due to acidic pH and urine waiting time (more than four hours). However, in the alkaline pH urine, the immunoassay results were not significantly altered, while GC-MS results were affected. For this reason, we can alkalinize urine to prevent false negative results in the immunoassay screening tests [7].

Peroxides: Peroxide-containing urine adulterants are highly effective for masking the presence of cocaine and opiates on screening and confirmatory assays. If we add four or five drops peroxide in urine, a dark brown color is performed due to the reduction of heptavalent chromium by hydrogen peroxide. Stealth is a combination of peroxide and peroxidase. The extent of successful concealment of opiates was inversely related to opiate concentration [8,9].

Pyridinium Chlorochromate (PCC): PCC was found to effectively produce false negative results for cocaine and amphetamine in Both Screening and Confirmatory Assays. PCC is commercially sold as Urine Luck, Klear II. The interference mechanism appears not to be a chemical change in the target drug but a decrease in standard pH levels of samples adulterated with PCC [9].

Glutaraldehyde: It was one of the earliest commercially adulterants, has sold as Clean X or Urine aid. These products cause interferences in immunoassay methods by decreasing absorbance rates, especially in cannabis tests. Additionally, glutaraldehyde didn't affect GC-MS results [10].

Non-oxidizing Adulterants

Table salt: Drug binding by changing protein structures [11]. Therefore, table salt affects the immunoassay results and its high doses produce false negative results. Studies have shown that sodium chloride increases urine density, which is one of the urine validation tests [12].

Detergent/Soap: These cleaning products are including surfactants and alkaline builders. Soap affects the drug binding on immunoassays by changing the pH of the urine sample. Soaps

and detergents cause false negative results for amphetamine, barbiturate, cannabis and cocaine [12].

Sodium Hydroxide: Drano drain cleaner is solid sodium hydroxide and it is consisting in variations of sodium hydroxide (lye), sodium hypochlorate (bleach), sodium nitrate, sodium chloride (salt) and aluminum. Sodium hydroxide is a caustic strong base. It causes change to alkaline pH in urine samples, then it effects the drug binding and solubility and produces false negative urine results [13].

Vinegar: Vinegar is a hydrous solution of acetic acid and called ethanoic acid or methane carboxylic acid. It decreases the urine pH levels, then effects the drug binding on immunoassay methods. Drug addicts use vinegar as a drug detox product. Vinegar disrupts the antigen-antibody association and effects especially amphetamine and cannabis results [11]. It was suggested that GC-MS confirmation test results aren't effected by non-oxidizing adulterants. Adding sodium hydrosulfide or sulfamic acid to the GC-MS method can help remove excess oxidizing additive and may prevent further oxidation of unchanged opium analytes in the sample, thus the accuracy of GC-MS results is less affected by oxidizing adulterants [13].

Conclusion

To prevent false negative results, it is important to apply the appropriate urine collection procedure. Not to forget the importance of surveillance while obtaining the sample because it is hard to detect the adulterants in analytical and post-analytical phase. Specimen integrity tests reduce false negative results. Also, in case of clinical suspicion, a validation test should be performed with the GC-MS method.

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