

Thermal Analysis on Acrylic Based Denture Materials

M Vasubabu¹, R Jeevan Kumar^{1*}, T Ramprasad^{1,2} and E Rekha¹

¹Department of Physics, Sri Krishnadevaraya University, India

²Department of Physics, MRITS College, India

Received: November 11, 2017; Published: December 12, 2017

*Corresponding author: R Jeevan Kumar, Molecular Bio-physical Laboratories, Department of Physics, Sri Krishnadevaraya University, Anantapur- 515 055, India, Email: rjkskuphy@gmail.com

Abstract

Acrylic is a polymer material made of poly methyl acrylate used as polymer electrolytes, complex polymer preparation and dental base material for fixation. Acrylic Polymeric materials reinforced with ceramic compositions provide advantages of high stiffness and strength to weight ratio as compared to conventional denture. The main objective of this work is to study variation in structural and physical and chemical transformations such as glass transition and degradations of Acrylic based denture materials such as Acralyn-H, RR Cold Cure material and Quick Ashvin were analyzed by X-ray diffraction, TGA and DTA.

Keywords: Acrylic based denture materials; X-rd TGA; DTA; Transition temperature; Degradations

Introduction

Acrylic resins were first utilized as denture base materials. Poly (methyl methacrylate) is the primary base resin used today. Polymerization of poly (methyl methacrylate) may be effected by heating the polymer-monomer mixture in a water bath, by chemical activation at room temperature using a material such as dimethyl-p-toluidine, or by activating the reaction using microwave energy or visible light [1]. Addition polymerization requires the activation of the initiator (benzoyl peroxide) to provide free radicals. Polymerization takes place as the free radicals open the double bonds of the methyl methacrylate, creating a chain reaction where the monomer attaches to polymer free radicals. Barron, Rueggeberg & Schuster [2] stated that the degree of monomer conversion of resin materials is a measure of the carbon double bonds (C=C) converted into carbon single bonds (C-C).

The goal of conventional/ rapid curing of acrylic resins is to completely polymerize the resin without porosity. In the conventional/ rapid curing methods, the monomer molecules are moved by thermal shocks from other molecules, and passively moved due to external heat. In the microwave method, the monomer molecules are moved by internal heat produced by a high-frequency electro-magnetic field [3-6]. This investigation involves structural and chemical changes in different resins by using thermal analysis.

Material and Methods

Three denture base resins namely Acralyn-H, RR Cold Cure material and Quick Ashvin were polymerized using casting method,

then samples are used characterized by x-Ray diffraction technique polymerization changes are analyzed by using TG-DTA.

Results and Discussion

The typical X-ray diffraction patterns were obtained in the present investigations for the various compositions as shown in the Fig.1 represents partially crystalline nature with peaks of three phases. The broad peak in the XRD pattern corresponds to the amorphous region of the material. The intensity of the crystalline peak same in all acrylic based resins. It clearly indicates that the crystallinity of acrylic based resin is due to thermoplastic phase. The acrylic based resin exhibits higher percentage shows more amorphous.

In the dental restorations, specific resin systems are applied to the damaged tooth area to form a cast restoration that is then heat-treated using special ovens under controlled laboratory conditions. The ideal temperature for heat treatment application depends on the thermal behavior of each composite, such as glass transition temperature (T_g) analysis and initial degradation temperature [7]. The T_g can successfully be used as a reference to sign the ideal heat treatment for photo-irradiated resin composites. Above T_g , the secondary molecular interactions are weakened and, as a consequence, material properties are optimized once trapped radicals are given the opportunity to react [8-9]. In addition, the maximum temperature for heating without damaging, i.e., initial degradation temperature, needs to be determined to avoid weight loss [10]. In the present investigation, Conventional TG-DTA is a

powerful and convenient thermal analysis technique which allows various important physical and chemical transformations such as glass transition and degradations are examined (Figures 1-4). Glass transition temperature (T_g) and number of phase transitions are evaluated for different dental materials are shown in Table 1.

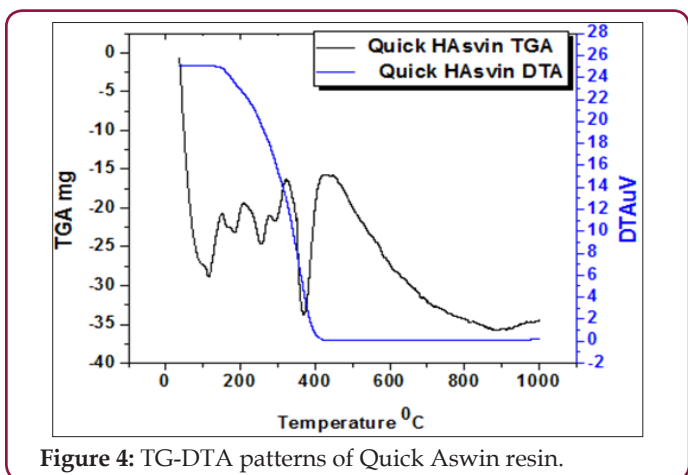
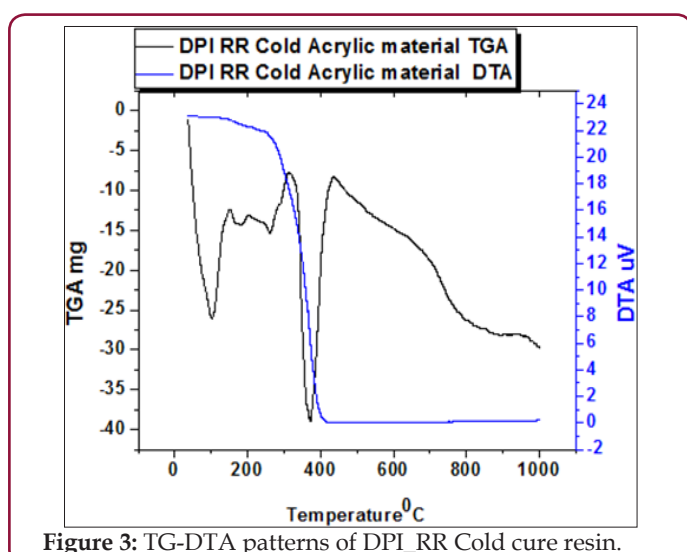
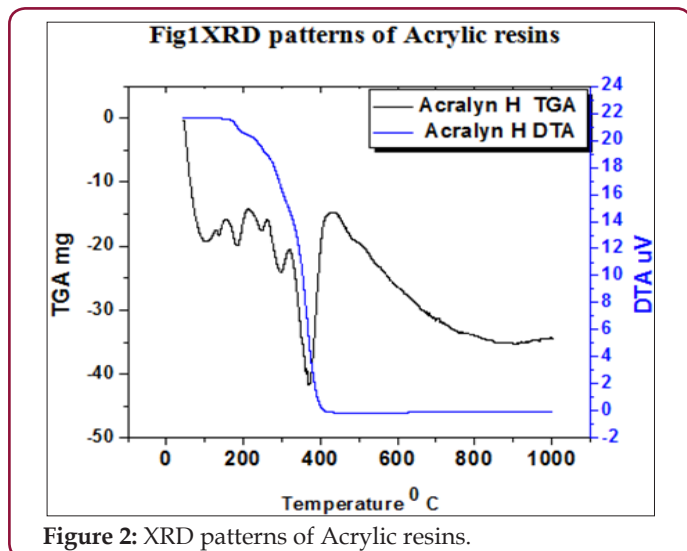
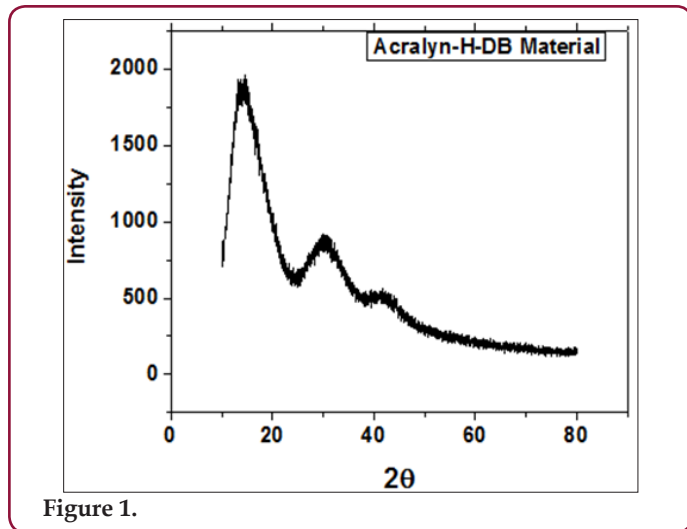


Table 1: Data on Thermal analysis of Acrylic resin.

Name of the Acrylic resin	Glass Transition Temperature(T_g) $^{\circ}$ C	No. Of. Phase Transition
Acralyn-H (AH)	103.43	3
RR Cold Cure material (RR)	101.73	2
Quick Ashvin (QA)	114.9	2

Conclusion

- a) This method of quantifying degree of polymerization represents a repeatable and expedient analysis of monomer conversion without the use of an internal standard.
- b) All resins obtained similar degrees of peaks determined by XRD.
- c) Acrylic resin such as Acralyn-H, RR Cold Cure material and Quick Ashvin values for the casting technique, though statistically significantly variation in weight percent with variable glass transition temperature (T_g).

References

1. Philips RW (1991) Skinners Science of Dental Materials, (9th edn). WB Saunders, Philadelphia, USA, pp. 177-213.
2. Buser D, Mericske-Stern R, Dula K, Lang NP (1999) Clinical experience with one-stage, non- submerged dental implants. Adv Dent Res 13: 153-159.
3. Al Doori D, Huggett R, Bates JF, Brooks SC (1988) A comparison of denture base acrylic resins polymerised by microwave irradiation and by conventional water bath curing systems. Dental Materials 4(1): 25-32.
4. Jerolimov V, Brooks SC, Huggett R, Bates JF (1989) Rapid curing of acrylic denture-base materials. Dent Materials 5(1): 18-25.
5. Yannikakis S, Zissis A, Polyzois, Andreopoulos A (2002) Evaluation of porosity in microwave- processed acrylic resin using photographic method. J Prosthet Dent 87(6): 613-619.
6. Compagnoni MA, Barbosa DB, De Souza RF, Pero AC (2004) the effect of polymerization cycles on porosity of microwave-processed denture base resin. J Prosthet Dent 91(3): 281-285.
7. Miyazaki CL, Medeiros IS, Santana IL, Matos JDR (2009) Rodrigues Filho LE, Heat treatment of a direct composite resin: influence on flexural strength. Braz Oral Res 23(3): 241-247.

8. Eldiwany M, Powers JM, George LA (1993) Mechanical properties of direct and post-cured composites. *Am J Dent* 6(5): 222-224.
9. Vaidhyanathan J, Vaidhya Nathan TK, wang Y, Viswanathan V (1992) Thermoanalytical characterization of visible light cure dental composites. *Journal of Oral Rehabilitation* 19(1): 49-64.
10. Santana IL, Lodovici E, Matos JR, Medeiros IS, Miyazaki CL, et al. (2009) Effect of experimental heat treatment on mechanical properties of resin composites. *Braz Dent Journal* 20(3): 205-210.



Assets of Publishing with us

- Global archiving of articles
- Immediate, unrestricted online access
- Rigorous Peer Review Process
- Authors Retain Copyrights
- Unique DOI for all articles

<http://biomedres.us/>